Quadruple hydrogen bonded oligo(*p***-phenylene vinylene) dimers**

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Received (in Cambridge, UK) 21st July 2000, Accepted 23rd August 2000 First published as an Advance Article on the web

Direct self-assembly of π -conjugated oligomers *via* self**complementary quadruple hydrogen bonding is achieved and the first steps towards supramolecular polymers with functional side chains are described.**

Conjugated polymers are investigated for a manifold of electronic semiconductor applications.¹ It is also widely accepted that well-defined oligomers will play a crucial role in the future advancement of π -conjugated materials, since their precise chemical structure and conjugation length stand for defined functional properties and facilitate enhanced control over their supramolecular architecture.2,3 Synthetic efforts have been mainly directed towards the synthesis of extended linear conjugated oligomers of length of 5–10 nm, but cyclic,4 bicyclic,⁵ spiro⁶ and dendritic⁷ structures have also been reported. Further, different π -conjugated functionalities have been combined in a large number of well-defined donor– acceptor molecules. The major challenge is now to obtain control of the supramolecular ordering over length scales extending into the macroscopic regime.8 Only a few examples have been reported on spatial organization of functional π conjugated entities by making use of hydrogen bonding motifs. Self-assembled fibers of mono and bisthiophene bisurea compounds have been reported and showed efficient charge transport within these fibers.⁹ Superstructure formation of π conjugated materials has also been obtained by hydrogen bonding complexation of perylene bisimide derivatives with a ditopic melamine compound10 and with chlorine singlet excited-state electron donor and a naphthalene bisimide acceptor.11

Recently, we have reported supramolecular polymers which exhibit real macroscopic polymeric properties.12 These polymeric systems are based on the strong dimerization of quadruple hydrogen-bonding self-complementary 2-ureido-4[1*H*]-ureido-

Scheme 1 Synthesis of **OPV3UP**. (a) di-tert-butyl tricarbonate, CH_2Cl_2 , 100%; (b) 6-tridecylisocytosine, pyridine, 90 °C, 26%.

pyrimidinone units. These recent developments pave the way to combine the specific electronic and optical properties of conjugated oligomers with the material properties of polymers, by incorporating well-defined π -conjugated moieties in these supramolecular polymeric assemblies. Here we report on the synthesis of the ureidopyrimidinone derivatives **OPV3UP**, **OPV4UP1** and **OPV4UP2** (depicted in Schemes 1 and 2) and on their supramolecular ordering.

Compound **OPV3UP** was prepared starting from the amino functionalized oligomer. Using di-*tert*-butyl tricarbonate the amine functionality was quantitatively converted to the isocyanate **2**.13 Reaction of **2** with 1.5 eq. of 6-tridecylisocytosine

Scheme 2 Synthesis of **OPV4UP1** and **OPV4UP2**. (a) diethyl 4-cyanobenzylphosphonate, *t*-BuOK, DMF, 91%; (b) LiAlH4, Et2O, 93%; (c) COCl2, toluene, 100 °C, 100%; (d) 6-tridecylisocytosine, pyridine, 90 °C, 52%; (e) diethyl 4-(methylbenzoate)phosphonate, *t*-BuOK, DMF, THF; (f) KOH, EtOH, 80 °C, 100%; (g) (COCl)₂, CH₂Cl₂, DMF, 100%; (h) MgCl₂, Et₃N, EtOAc, HCl aq., 30%; (i) (NH₂)₂C=NH·HCO₃, EtOH, 90%; (j) n-C₄H₉NCO, pyridine, 90 °C.

Fig. 1¹H-NMR spectra of **OPV4UP1** recorded in CDCl₃.

in anhydrous pyridine at 90 °C afforded **OPV3UP** in 26% yield after column chromatography. The low yield of **OPV3UP** is probably due to the relatively low reactivity of aromatic isocyanates. Therefore, benzylic isocyanates were used for the preparation of **OPV4UP1**. To obtain the benzylic isocyanate, aldehyde **3** was allowed to react with diethyl 4-cyanobenzylphosphonate in a Wittig–Horner reaction to give the nitrile compound in 91% yield. The nitrile functionality was subsequently reduced using LAH to afford the benzylic amine. The pure compound was isolated after work-up and precipitation in 93% yield. The amine derivative was reacted with phosgene in refluxing toluene to give the isocyanate. Reaction of the latter with 6-tridecylisocytosine in anhydrous pyridine at 90 °C gave **OPV4UP1** in 52% yield. In order to obtain full conjugation between the OPV and hydrogen bonding unit, **OPV4UP2** was synthesized. Reaction of aldehyde **3** with diethyl 4-(methylbenzoate)phosphonate in a Wittig–Horner reaction afforded a mixture of esters **5**. Saponification of these esters gave quantitatively the acid, which was quantitatively converted to the acid chloride using oxalylchloride in DCM and DMF. The β -ketoester was obtained in 30% yield by reaction of the acid chloride with potassium ethylmalonate in presence of triethylamine and magnesium chloride. Reaction of the β -ketoester with guanidinium carbonate yielded quantitatively the isocytosine. The desired compound **OPV4UP2** was finally obtained in 90% yield by reaction of the isocytosine and *n*-butylisocyanate in anhydrous pyridine at 90 °C.

All the compounds **OPV3UP**, **OPV4UP1** and **OPV4UP2** were fully characterized.† These π -conjugated oligomers form quadruple hydrogen bonded **DDAA**-dimers in solution as is evident from the ¹H-NMR spectra (Fig. 1). The large downfield shift for the N–H protons gives direct evidence for the involvement of these protons in strong hydrogen bonding.14 The electronic absorption spectra recorded in $CHCl₃$ solution display a weak band in the visible spectral region ($\lambda_{\text{max}} = 409$) nm) for the less extended π -conjugated **OPV3UP**. In the case of tetrameric oligophenylene vinylene, **OPV4UP1** and **OPV4UP2**, the absorption maxima are located at $\lambda_{\text{max}} = 431$ and 446 nm, respectively. The red shift of the absorption maxima of **OPV4UP2** indicates conjugation between the OPV segment and the hydrogen bonding unit. In addition, in dilute CHCl₃ solution (*ca.* 10^{-5} M) these hydrogen bonded species are present in dimeric form since the association constant of the ureidopyrimidinone units is extremely high ($K_{\text{dim}} = 6 \times 10^7$) M^{-1} in CHCl₃ solutions).¹⁵ In dodecane solution, both the dimers of **OPV4UP1** and **OPV4UP2** aggregate into larger architectures and circular dichroism (CD) spectroscopy shows an exciton coupling of the $\pi-\pi^*$ transition in this solvent. In other words, the side-chain chirality is expressed at the supramolecular level.¹⁶ In the bulk, the two trialkoxybenzene capped dimers are liquid crystalline, but investigation concerning their detailed structure is in progress.

In conclusion, directed self-assembly of π -conjugated oligomers *via* self-complementary quadruple hydrogen bonding has been achieved. It opens new possibilities for the design of electronically active supramolecular materials in which the specific properties of well-defined oligomers can be combined with the material properties of polymers.

We thank Michel Fransen for the synthesis of the starting materials, Joost van Dongen for MALDI-TOF MS measurements, and Dr Rint Sijbesma for fruitful and helpful discussions. This work has been supported by Netherlands Organization for Scientific Research (NWO) and the Royal Netherlands Academy of Arts and Sciences.

Notes and references

† Full synthetic details will be given elsewhere. All new compounds were authenticated by ¹H and ¹³C NMR, FT-IR, MALDI-TOF MS and elemental analyses. *Selected data*: for **OPV3UP**: δ_{N-H} (CDCl₃) 12.91, 12.35, 12.11, MALDI-TOF MS (MW = 974.68) $m/z = 974.77$ [M]⁺, elemental analyses C, 74.46 (75.11), H, 9.28 (9.30), N, 5.96 (5.74%); **OPV4UP1**: δ_{N-H} (CDCl₃) 13.06, 12.07, 10.89, MALDI-TOF MS (MW = 1629.28) m/z = 1629.89 [M]+, elemental analyses C, 77.68 (77.35), H, 10.77 (10.39), N, 3.26 (3.44%); **OPV4UP2**: δ_{N-H} (CDCl₃) 13.91, 12.03, 10.29, MALDI-TOF MS (MW = 1490.24) *m/z* = 1490.09 [M]+, elemental analyses C, 75.99 (76.60), H, 9.58 (10.00), N, 3.45 (3.80%); calculated values in parenthesis.

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