Quadruple hydrogen bonded oligo(p-phenylene vinylene) dimers

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Direct self-assembly of π -conjugated oligomers *via* selfcomplementary quadruple hydrogen bonding is achieved and the first steps towards supramolecular polymers with functional side chains are described.

Conjugated polymers are investigated for a manifold of electronic semiconductor applications.¹ It is also widely accepted that well-defined oligomers will play a crucial role in the future advancement of π -conjugated materials, since their precise chemical structure and conjugation length stand for defined functional properties and facilitate enhanced control over their supramolecular architecture.^{2,3} Synthetic efforts have been mainly directed towards the synthesis of extended linear conjugated oligomers of length of 5-10 nm, but cyclic,⁴ bicyclic,⁵ spiro⁶ and dendritic⁷ structures have also been reported. Further, different π -conjugated functionalities have been combined in a large number of well-defined donoracceptor molecules. The major challenge is now to obtain control of the supramolecular ordering over length scales extending into the macroscopic regime.⁸ Only a few examples have been reported on spatial organization of functional π conjugated entities by making use of hydrogen bonding motifs. Self-assembled fibers of mono and bisthiophene bisurea compounds have been reported and showed efficient charge transport within these fibers.⁹ Superstructure formation of π conjugated materials has also been obtained by hydrogen bonding complexation of perylene bisimide derivatives with a ditopic melamine compound¹⁰ and with chlorine singlet excited-state electron donor and a naphthalene bisimide acceptor.11

Recently, we have reported supramolecular polymers which exhibit real macroscopic polymeric properties.¹² These polymeric systems are based on the strong dimerization of quadruple hydrogen-bonding self-complementary 2-ureido-4[1*H*]-ureido-



Scheme 1 Synthesis of OPV3UP. (a) di-*tert*-butyl tricarbonate, CH_2Cl_2 , 100%; (b) 6-tridecylisocytosine, pyridine, 90 °C, 26%.

pyrimidinone units. These recent developments pave the way to combine the specific electronic and optical properties of conjugated oligomers with the material properties of polymers, by incorporating well-defined π -conjugated moieties in these supramolecular polymeric assemblies. Here we report on the synthesis of the ureidopyrimidinone derivatives **OPV3UP**, **OPV4UP1** and **OPV4UP2** (depicted in Schemes 1 and 2) and on their supramolecular ordering.

Compound **OPV3UP** was prepared starting from the amino functionalized oligomer. Using di-*tert*-butyl tricarbonate the amine functionality was quantitatively converted to the iso-cyanate $2^{.13}$ Reaction of 2 with 1.5 eq. of 6-tridecylisocytosine



Scheme 2 Synthesis of OPV4UP1 and OPV4UP2. (a) diethyl 4-cyanobenzylphosphonate, *t*-BuOK, DMF, 91%; (b) LiAlH₄, Et₂O, 93%; (c) COCl₂, toluene, 100 °C, 100%; (d) 6-tridecylisocytosine, pyridine, 90 °C, 52%; (e) diethyl 4-(methylbenzoate)phosphonate, *t*-BuOK, DMF, THF; (f) KOH, EtOH, 80 °C, 100%; (g) (COCl₂, CH₂Cl₂, DMF, 100%; (h) MgCl₂, Et₃N, EtOAc, HCl aq., 30%; (i) (NH₂)₂C=NH·HCO₃, EtOH, 90%; (j) n-C₄H₉NCO, pyridine, 90 °C.

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Fig. 1 ¹H-NMR spectra of OPV4UP1 recorded in CDCl₃.

in anhydrous pyridine at 90 °C afforded OPV3UP in 26% yield after column chromatography. The low yield of OPV3UP is probably due to the relatively low reactivity of aromatic isocyanates. Therefore, benzylic isocyanates were used for the preparation of **OPV4UP1**. To obtain the benzylic isocyanate, aldehyde 3 was allowed to react with diethyl 4-cyanobenzylphosphonate in a Wittig-Horner reaction to give the nitrile compound in 91% yield. The nitrile functionality was subsequently reduced using LAH to afford the benzylic amine. The pure compound was isolated after work-up and precipitation in 93% yield. The amine derivative was reacted with phosgene in refluxing toluene to give the isocyanate. Reaction of the latter with 6-tridecylisocytosine in anhydrous pyridine at 90 °C gave OPV4UP1 in 52% yield. In order to obtain full conjugation between the OPV and hydrogen bonding unit, OPV4UP2 was synthesized. Reaction of aldehyde 3 with diethyl 4-(methylbenzoate)phosphonate in a Wittig-Horner reaction afforded a mixture of esters 5. Saponification of these esters gave quantitatively the acid, which was quantitatively converted to the acid chloride using oxalylchloride in DCM and DMF. The β -ketoester was obtained in 30% yield by reaction of the acid chloride with potassium ethylmalonate in presence of triethylamine and magnesium chloride. Reaction of the β -ketoester with guanidinium carbonate yielded quantitatively the isocytosine. The desired compound **OPV4UP2** was finally obtained in 90% yield by reaction of the isocytosine and *n*-butylisocyanate in anhydrous pyridine at 90 °C

All the compounds OPV3UP, OPV4UP1 and OPV4UP2 were fully characterized.[†] These π -conjugated oligomers form quadruple hydrogen bonded DDAA-dimers in solution as is evident from the ¹H-NMR spectra (Fig. 1). The large downfield shift for the N-H protons gives direct evidence for the involvement of these protons in strong hydrogen bonding.14 The electronic absorption spectra recorded in CHCl₃ solution display a weak band in the visible spectral region ($\lambda_{max} = 409$ nm) for the less extended π -conjugated **OPV3UP**. In the case of vinylene, **OPV4UP1** tetrameric oligophenylene and **OPV4UP2**, the absorption maxima are located at $\lambda_{max} = 431$ and 446 nm, respectively. The red shift of the absorption maxima of **OPV4UP2** indicates conjugation between the OPV segment and the hydrogen bonding unit. In addition, in dilute CHCl₃ solution (ca. 10^{-5} M) these hydrogen bonded species are present in dimeric form since the association constant of the ureidopyrimidinone units is extremely high ($K_{\rm dim} = 6 \times 10^7$ M^{-1} in CHCl₃ solutions).¹⁵ In dodecane solution, both the dimers of OPV4UP1 and OPV4UP2 aggregate into larger architectures and circular dichroism (CD) spectroscopy shows an exciton coupling of the π - π * transition in this solvent. In other words, the side-chain chirality is expressed at the supramolecular level.¹⁶ In the bulk, the two trialkoxybenzene capped dimers are liquid crystalline, but investigation concerning their detailed structure is in progress.

In conclusion, directed self-assembly of π -conjugated oligomers *via* self-complementary quadruple hydrogen bonding has been achieved. It opens new possibilities for the design of electronically active supramolecular materials in which the specific properties of well-defined oligomers can be combined with the material properties of polymers.

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Notes and references

† Full synthetic details will be given elsewhere. All new compounds were authenticated by ¹H and ¹³C NMR, FT-IR, MALDI-TOF MS and elemental analyses. *Selected data*: for **OPV3UP**: δ_{N-H} (CDCl₃) 12.91, 12.35, 12.11, MALDI-TOF MS (MW = 974.68) m/z = 974.77 [M]⁺, elemental analyses C, 74.46 (75.11), H, 9.28 (9.30), N, 5.96 (5.74%); **OPV4UP1**: δ_{N-H} (CDCl₃) 13.06, 12.07, 10.89, MALDI-TOF MS (MW = 1629.28) m/z = 1629.89 [M]⁺, elemental analyses C, 77.68 (77.35), H, 10.77 (10.39), N, 3.26 (3.44%); **OPV4UP2**: δ_{N-H} (CDCl₃) 13.91, 12.03, 10.29, MALDI-TOF MS (MW = 1490.24) m/z = 1490.09 [M]⁺, elemental analyses C, 75.99 (76.60), H, 9.58 (10.00), N, 3.45 (3.80%); calculated values in parenthesis.

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